



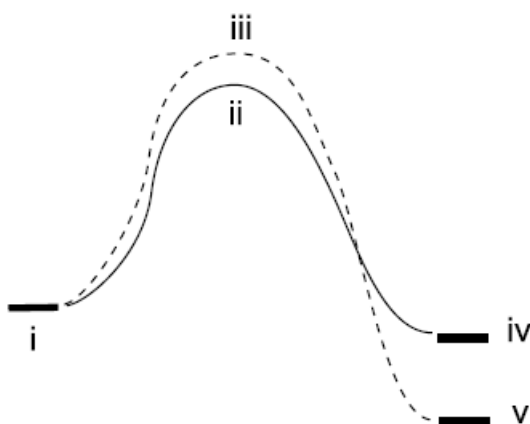
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Consider the energy level diagram shown below which represents two competing molecular rearrangements. Which of the statements is false?



- ☐ (A) The energy difference between species i and species ii represents the energy of activation for the formation of species iv.
- ☐ (B) Species iv is the kinetic product.
- ☐ (C) Species v is the thermodynamic product
- ☐ (D) Carrying out the reaction at higher temperature will likely produce a greater percentage of species v.
- ☐ (E) Species iii represents the transition state leading to compound v.
- ☐ (F) Lowering the energy of species iii will lead to a higher concentration of v at equilibrium

VIDEO SOLUTION



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