



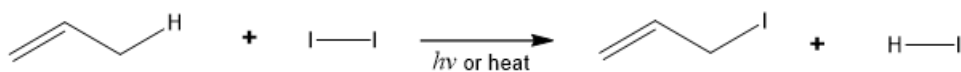
ORGOREVIEW

Question Vault

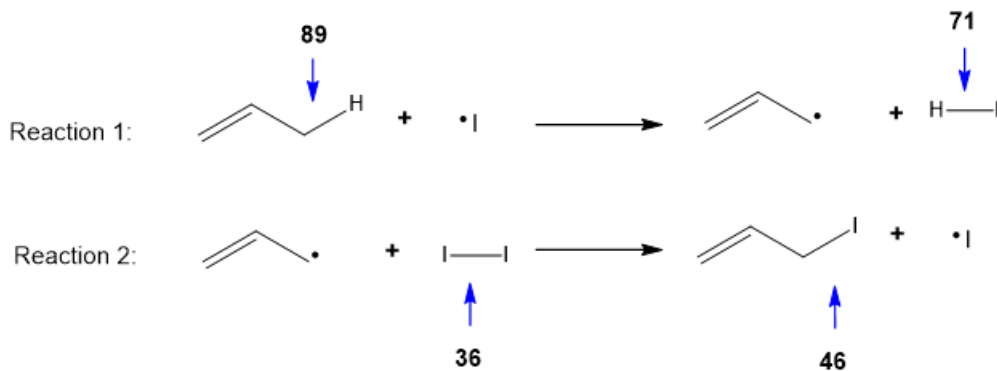
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Professor Lee proposed the following reaction for the preparation of 3-iodo-1-propene from propene.



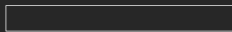
The following two propagation steps were hypothesized. The relevant bond energies in kcal/mole are shown.



- ☐ (A) The first reaction is endothermic. ☐ (B) The second reaction is exothermic.
☐ (C) The overall reaction is exothermic. ☐ (D) The $\text{I}-\text{I}$ bond is weaker than the $\text{H}-\text{I}$ bond.
☐ (E) The proposed reaction is not a good method for the preparation of 3-iodo-1-propene from propene.

VIDEO SOLUTION





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