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Choose the incorrect statement about reactions of haloalkanes proceeding by S_N2 mechanisms.

- \circ (A) The order of reactivity of the haloalkane is the following: methyl (fastest) > 1° > 2° > 3°
- © (B) Rearrangements are common. © (C) Inversion of configuration occurs at the site of substitution.
- \circ (D) The reaction rate depends upon the concentration of the nucleophile and haloalkane.
- © (E) The reaction rate is generally faster in polar aprotic solvents.

How to Reach Us Todd's Test Prep 2255 Glades Road Suite 324A Boca Raton, Florida 33431 E-mail: help@orgoreview.com Links Contact Terms Privacy Policy

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