



# ORGOREVIEW

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Choose the incorrect statement about reactions of haloalkanes proceeding by  $S_N2$  mechanisms.

- (A) The order of reactivity of the haloalkane is the following: methyl (fastest)  $> 1^\circ > 2^\circ > 3^\circ$
- (B) Rearrangements are common.      (C) Inversion of configuration occurs at the site of substitution.
- (D) The reaction rate depends upon the concentration of the nucleophile and haloalkane.
- (E) The reaction rate is generally faster in polar aprotic solvents.

### How to Reach Us

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