



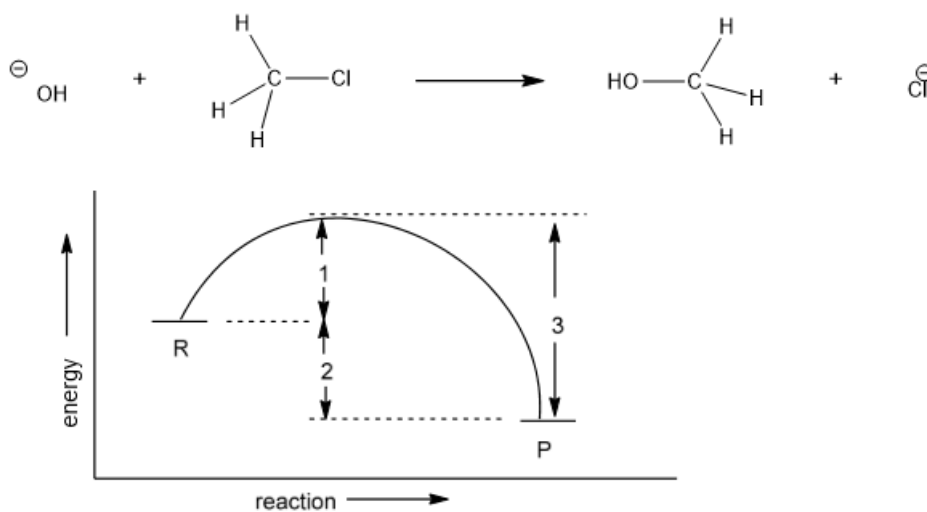
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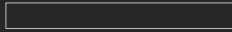


- i. Energy difference 1 is related to the rate of the reaction.
- ii. Energy difference 2 is related to equilibrium constant.
- iii. If the energy of the reaction (R) is decreased, the reaction to go faster.

(A) i (B) ii (C) iii (D) i + ii (E) i + iii (F) ii + iii

VIDEO SOLUTION





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