

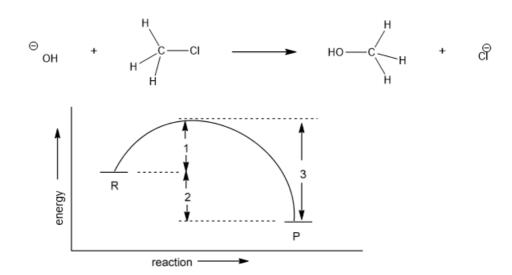
## **ORGOREVIEW**

## Question Vault

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Choose the true statements about the following reaction profile.



- i. Energy difference 1 is related to the rate of the reaction.
- ii. Energy difference 2 is related to equilibrium constant.
- iii. If the energy of the reaction (R) is decreased, the reaction to go faster.
- O (A) i
- **(B)** ii

- $\circ$  (C) iii  $\circ$  (D) i + ii  $\circ$  (E) i + iii  $\circ$  (F) ii + iii

**VIDEO SOLUTION** 

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