



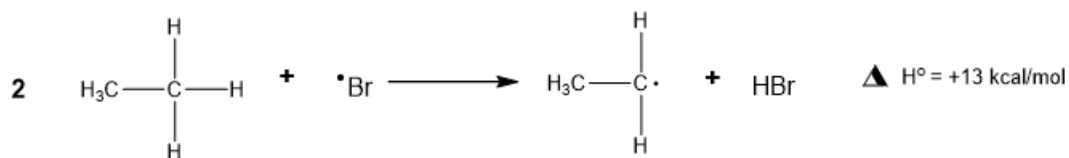
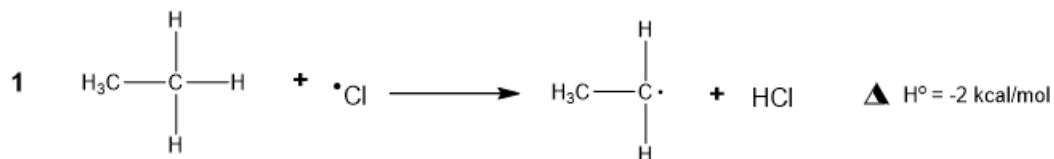
# ORGOREVIEW

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Choose the **incorrect** statement about the following two reactions.



- ☐ (A) Reaction (1) would be expected to be faster than reaction (2).
- ☐ (B) According to Hammond's postulate the transition state of reaction (2) resembles the product.
- ☐ (C) According to Hammond's postulate the transition state of reaction (1) resembles the reactant.
- ☐ (D) The C-H bond of reaction (2) is completely broken in the transition state.
- ☐ (E) The reaction of ethane with the fluorine radical ( $\cdot\text{F}$ ) would be expected to be faster than either (1) or (2)

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