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Because of the higher bond strength of H-F (570 kJ/mol) compared to H-Cl (432 kJ/mol), the first step in fluorination is more exothermic than the first step in chlorination. Application of Hammond's postulate to these reactions predicts that fluorination will be more selective than chlorination.

- ☐ (A) Very sure this is true ☐ (B) Somewhat sure this is true ☐ (C) Very sure this is false
☐ (D) Somewhat sure this is false

VIDEO SOLUTION



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