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Which one of the following statements is not true regarding resonance structures of a given molecule?

- ☐ (A) All resonance structures must have the same number of electrons.
- ☐ (B) All resonance structures must have the same arrangement of atoms.
- ☐ (C) Each atom in all of the resonance structures must have a complete shell of valence electrons.
- ☐ (D) All resonance structures must be valid Lewis structures.
- ☐ (E) The resonance structures may have different energies.

VIDEO SOLUTION



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