



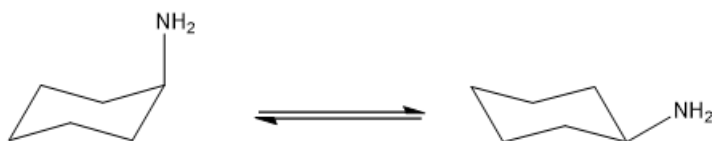
ORGOREVIEW

Question Vault

Welcome to orgoreview question vault. We have 5000 more problems for you to solve.

Share

Estimate the equilibrium constant for the following transformation.



$$\begin{aligned}\Delta G^\circ &= -RT \ln K_{eq} \\ &= -5.72 \text{ J mol}^{-1} \log K_{eq} \text{ (at room temperature)}\end{aligned}$$

- ☐ (A) 0.01 ☐ (B) 0.1 ☐ (C) 1 ☐ (D) 10 ☐ (E) 100

How to Reach Us

Todd's Test Prep
2255 Glades Road
Suite 324A
Boca Raton, Florida 33431
E-mail: help@orgoreview.com

Links

[Contact](#)

[Terms](#)

[Privacy Policy](#)

follow us



Copyright 2026. All Rights Reserved.