



# ORGOREVIEW

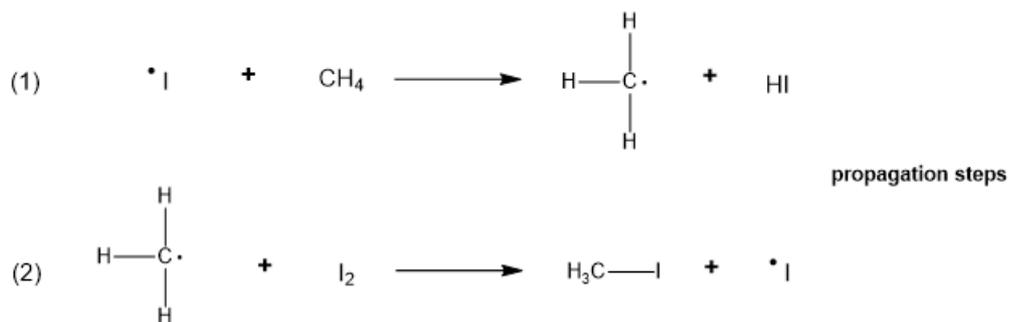
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Choose the correct statement(s) for the two propagating steps for the iodination of methane.

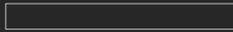


- (a) The concentration of  $\text{CH}_4$  and  $\text{I}$  radical are equal in reaction mixture.  
(b) Step (1) is slower than step (2)  
(c) The free radical chain reaction will be terminated by reaction of  $\cdot\text{CH}_3$  and  $\cdot\text{I}$  radical

(A) (a) + (b)    (B) (a) + (c)    (C) (b) + (c)    (D) (a)    (E) (b)    (F) (c)

VIDEO SOLUTION





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