



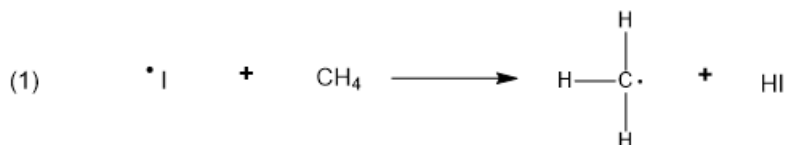
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Question Vault

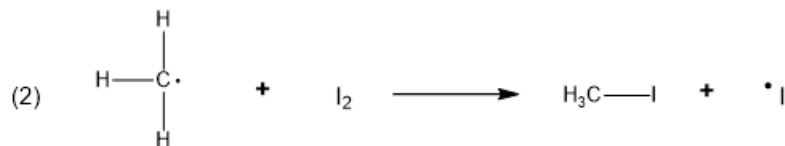
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Choose the correct statement(s) for the two propagating steps for the iodination of methane.



propagation steps



- (a) The concentration of CH_4 and I radical are equal in reaction mixture.
(b) Step (1) is slower than step (2)
(c) The free radical chain reaction will be terminated by reaction of $\cdot\text{CH}_3$ and $\cdot\text{I}$ radical

☐ (A) (a) + (b) ☐ (B) (a) + (c) ☐ (C) (b) + (c) ☐ (D) (a) ☐ (E) (b) ☐ (F) (c)

VIDEO SOLUTION



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