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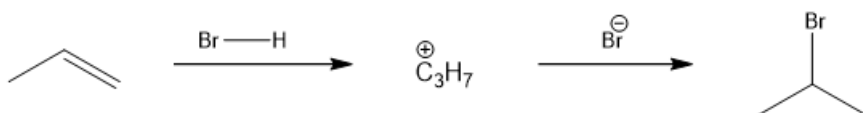
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Propene will react with H-Br to give 2-bromopropane. The reaction occurs in two steps. First the propene reacts with H-Br to give a cationic intermediate. Then the cationic intermediate reacts with Br⁻ to give the product.

Draw a curved arrow mechanism for the reaction. Clearly show all bond breaking and bond making steps. Include the structure of the cationic intermediate.



VIDEO SOLUTION



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