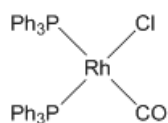# ORGOREVIEW

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Which statement is true regarding the rhodium complex shown?

- ○ **(A)** The oxidation state of rhodium is 0 and the number of valence electrons is 16.
- ○ **(B)** The oxidation state of rhodium is 0 and the number of valence electrons is 17.
- ○ **(C)** The oxidation state of rhodium is +1 and the number of valence electrons is 16.
- ○ **(D)** The oxidation state of rhodium is +1 and the number of valence electrons is 16.
- ○ **(E)** The oxidation state of rhodium is +2 and the number of valence electrons is 15.
- ○ **(F)** The oxidation state of rhodium is +2 and the number of valence electrons is 16.

VIDEO SOLUTION

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