



# ORGOREVIEW

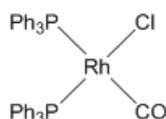
## Question Vault

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Which statement is true regarding the rhodium complex shown?



- (A) The oxidation state of rhodium is 0 and the number of valence electrons is 16.
- (B) The oxidation state of rhodium is 0 and the number of valence electrons is 17.
- (C) The oxidation state of rhodium is +1 and the number of valence electrons is 16.
- (D) The oxidation state of rhodium is +1 and the number of valence electrons is 16.
- (E) The oxidation state of rhodium is +2 and the number of valence electrons is 15.
- (F) The oxidation state of rhodium is +2 and the number of valence electrons is 16.

## VIDEO SOLUTION



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