

## **ORGOREVIEW**

## Question Vault

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Choose the answer with the most correct statements about the following acid-base reaction.

$$\mathsf{H} \underline{\hspace{0.5cm}} \mathsf{C} \underline{\hspace{0.5cm}} \mathsf{C} \underline{\hspace{0.5cm}} \mathsf{H} \quad + \quad \stackrel{\ominus}{:} \overset{\ominus}{\mathsf{N}} \mathsf{H}_2 \qquad \qquad \bullet \qquad \qquad - \mathsf{C} \underline{\overset{\ominus}{\hspace{0.5cm}}} \mathrel{\overset{\bigcirc}{:}} : \quad + \quad : \mathsf{N} \mathsf{H}_3$$

- 1. The strongest acid in this reaction is H—C ≡ C—H
- 2. The strongest base in this reaction is H—C  $\stackrel{\Theta}{=}$  C:
- 3. H—C  $\stackrel{\Theta}{==}$  : is the conjugate base of H—C  $\stackrel{\Theta}{==}$  C—H
- 4. : NH<sub>3</sub> is the conjugate acid of : NH<sub>2</sub>

## **VIDEO SOLUTION**

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