

## **ORGOREVIEW**

## **Question Vault**

Welcome to orgoreview question vault. We have 5000 more problems for you to solve.

Share

Which of the following is a true statement regarding kinetic vs. thermodynamic control?

- C (A) When the products are in equilibrium under the reaction conditions, the relative amounts of products depend on their stabilities.
- © (B) When a reaction is under thermodynamic control, the relative amounts of products depends on the activation energies of the steps leading to their formation.
- C (C) The kinetic product always predominates when the reaction is reversible.
- © (D) Higher temperatures and longer reaction times tend to favor the kinetic product over the thermodynamic product.
- (E) None of the above

**VIDEO SOLUTION** 

Ö

How to Reach Us
Todd's Test Prep
2255 Glades Road
Suite 324A
Boca Raton, Florida 33431
E-mail: help@orgoreview.com

	Links
	<u>Contact</u>
	<u>Terms</u>
	Privacy Policy
	follow us
	Copyright 2025. All Rights Reserved.
<del>-</del>	