



ORGOREVIEW

Question Vault

Welcome to orgoreview question vault. We have 5000 more problems for you to solve.

Share

Which of the following is a *true* statement regarding kinetic vs. thermodynamic control?

- (A) When the products are in equilibrium under the reaction conditions, the relative amounts of products depend on their stabilities.
- (B) When a reaction is under thermodynamic control, the relative amounts of products depends on the activation energies of the steps leading to their formation.
- (C) The kinetic product always predominates when the reaction is reversible.
- (D) Higher temperatures and longer reaction times tend to favor the kinetic product over the thermodynamic product.
- (E) None of the above

VIDEO SOLUTION



How to Reach Us

Todd's Test Prep
2255 Glades Road
Suite 324A

Boca Raton, Florida 33431

E-mail: help@orgoreview.com

Links

[Contact](#)

[Terms](#)

[Privacy Policy](#)

follow us



Copyright 2026. All Rights Reserved.