



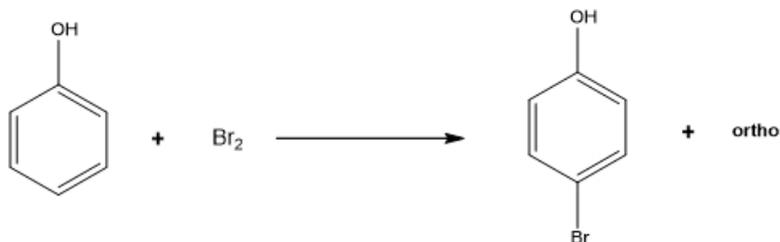
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Phenol is sufficiently reactive in electrophilic aromatic substitution that it undergoes bromination even in the absence of a Lewis acid catalyst. Draw the structure of the arenium ion intermediate, $[\text{C}_6\text{H}_6\text{BrO}]^+$, in bromination at the para position and draw all of the important resonance structures for this intermediate.



VIDEO SOLUTION



How to Reach Us

Todd's Test Prep
2255 Glades Road
Suite 324A
Boca Raton, Florida 33431

E-mail: help@orgoreview.com

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