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The following bromination of 2, 4-hexadiene using only one equivalent of Br2 will yield two major products, each with molecular formula $C_6H_{10}Br_2$. The two products are constitutional isomers. One of the products is shown below. Draw the structure of the other major products. Note that each major product is a mixture of stereoisomers.

VIDEO SOLUTION

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How to Reach Us

Todd's Test Prep

2255 Glades Road

Suite 324A

Boca Raton, Florida 33431

E-mail: help@orgoreview.com

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