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Question Vault

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Carbonyl compounds can form a hydrate as shown in the following equilibrium equation. Which of the compounds would be predicted to have the highest equilibrium constant for hydrate formation?

$$K_{eq} = ?$$

$$C (A) \qquad C (B) \qquad C (C) \qquad C (D) \qquad C (E)$$

$$H_{3}C \qquad CH_{3} \qquad F_{3}C \qquad CH_{3} \qquad F_{3}C \qquad CF_{3} \qquad H_{3}C \qquad CH_{3}$$

VIDEO SOLUTION

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