

ORGOREVIEW

Question Vault

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Consider the bromination of methane.

- C (A) Reaction (1) is called the chain initiation step. C (B) Reaction (4) is called the chain termination step.
- \bigcirc **(C)** The \triangle H° for step (2) is +71 kJ/mol. \bigcirc **(D)** The \triangle H° for step (3) is -101 kJ/mol.
- \circ (E) The ΔH° for overall reaction is -172 kJ/mol.

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