



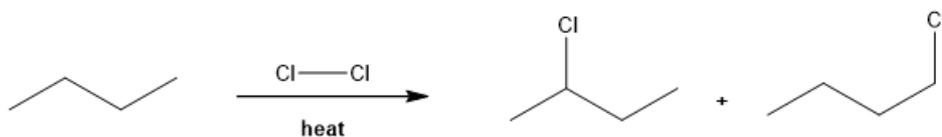
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Choose the correct statements about the monochlorination of butane. (note: 2° C-H bonds are two times more reactive than 1° C-H bonds in chlorination reactions.)



- (a) The major product of this reaction is 2-chlorobutane
(b) The product mixture will rotate the plane of polarized light
(c) Under the same conditions as chlorination, bromination of butane would occur at a slow rate
- (A) (a) + (b) (B) (a) + (c) (C) (b) + (c) (D) (a) (E) (b) (F) (c)

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